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seawater is mostly  $H_2O$ . But about 3.5% of seawater is dissolved solids, mostly NaCl but other ions as well.

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solution, but you can add more sugar if you like, and you still have a solution. By contrast, compounds are also composed of two or more substances (usually elements) but in a ratio that cannot vary.

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1. Explain why solutions are classified as mixtures instead of compounds. 2.

Solutions are \_\_\_\_\_ mixtures made up of very small particles that are actually molecules, \_\_\_\_\_, or \_\_\_\_\_. 3. Solutions are said to be in a \_\_\_\_\_ phase even though the components may have been

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in different phases before the solution was formed. 4.

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added to a solution of 5.77 g of AgNO<sub>3</sub>?  
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Answers A solution is a homogeneous mixture —a mixture of two or more substances that are so intimately mixed that the mixture behaves in many ways like a single substance.

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